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DESCRIPTORS: "Hydrocarbons, Oxidation," "Ethane,
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In the oxidation of ethane in stoichiometric proportions
with oxygen the reaction rate at a temperature of 466
and 486°C depends upon the quantity of the substance
that has reacted and upon the quantity of water vapour
added. It is impossible to explain the acceleration of
the reaction by the catalytic action of water alone,
some kind of intermediate products exist that cause the
(Chemistry--Organic, TT, v. 6, no. 7) (over)

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Below 600° water vapor catalyzes oxidation of ethane,
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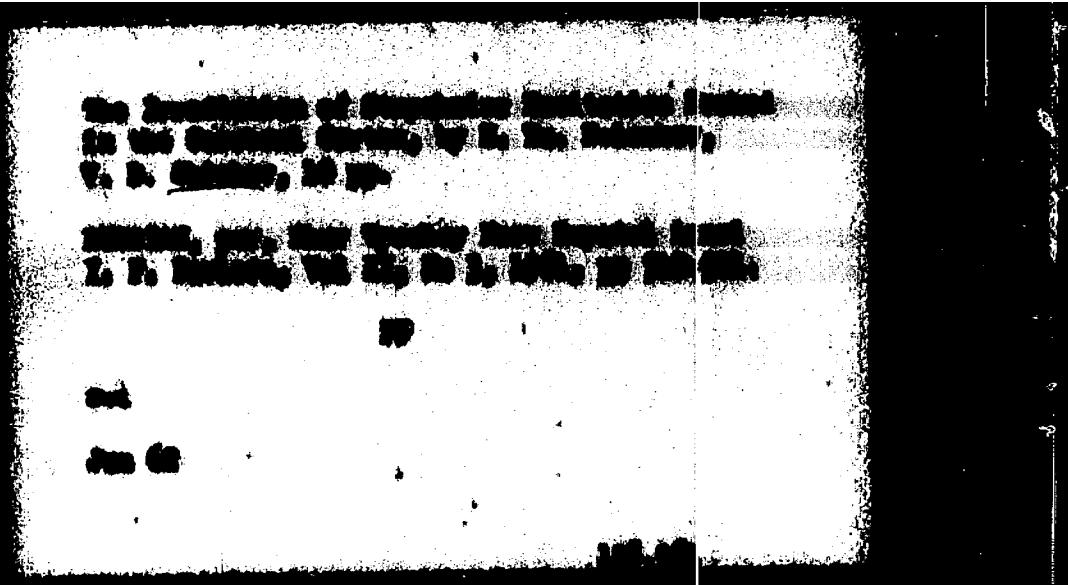
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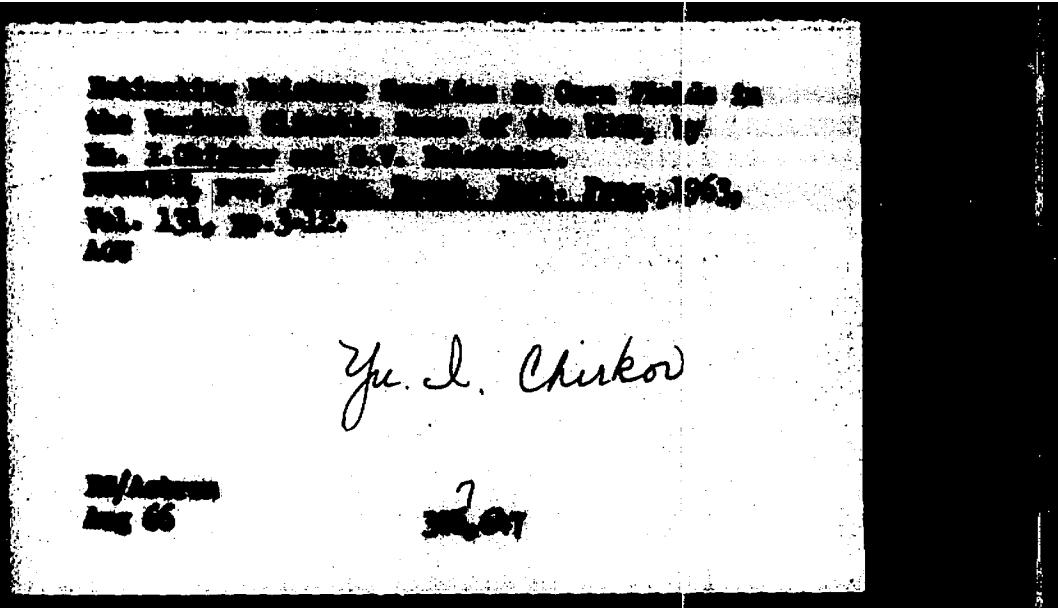
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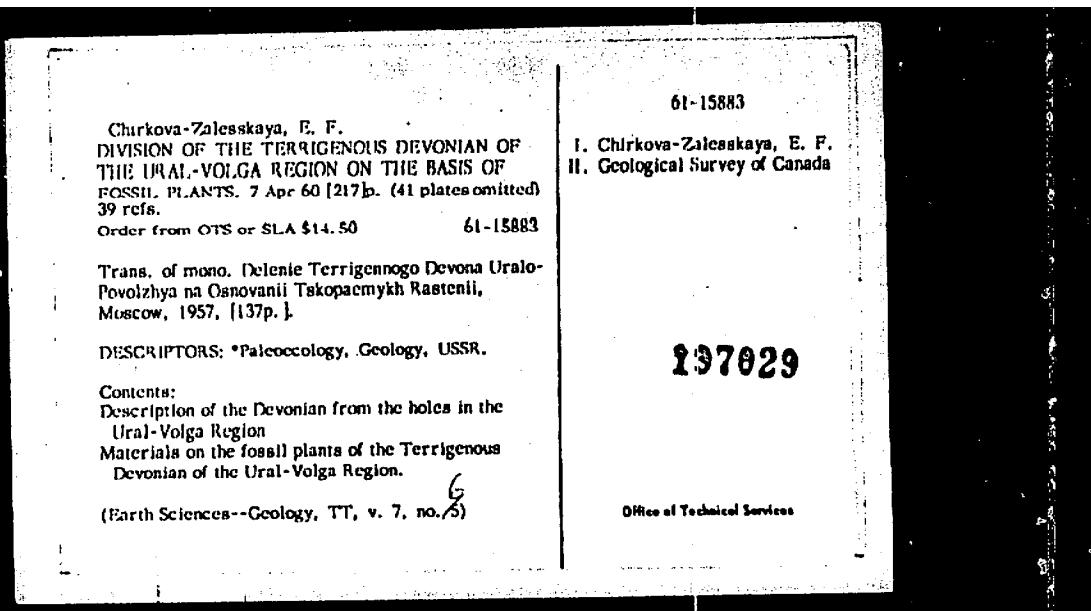
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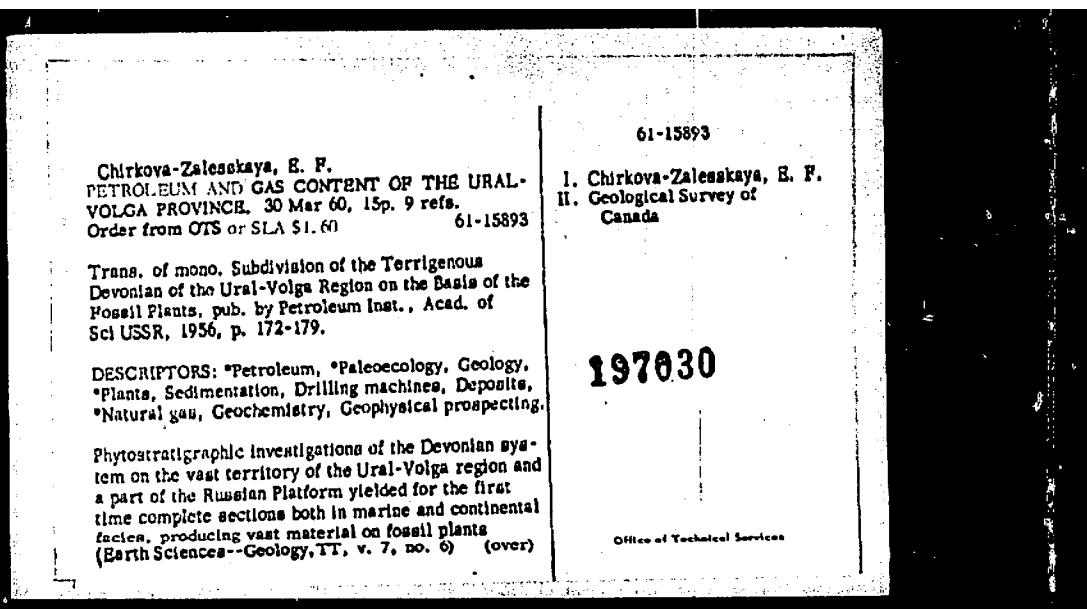
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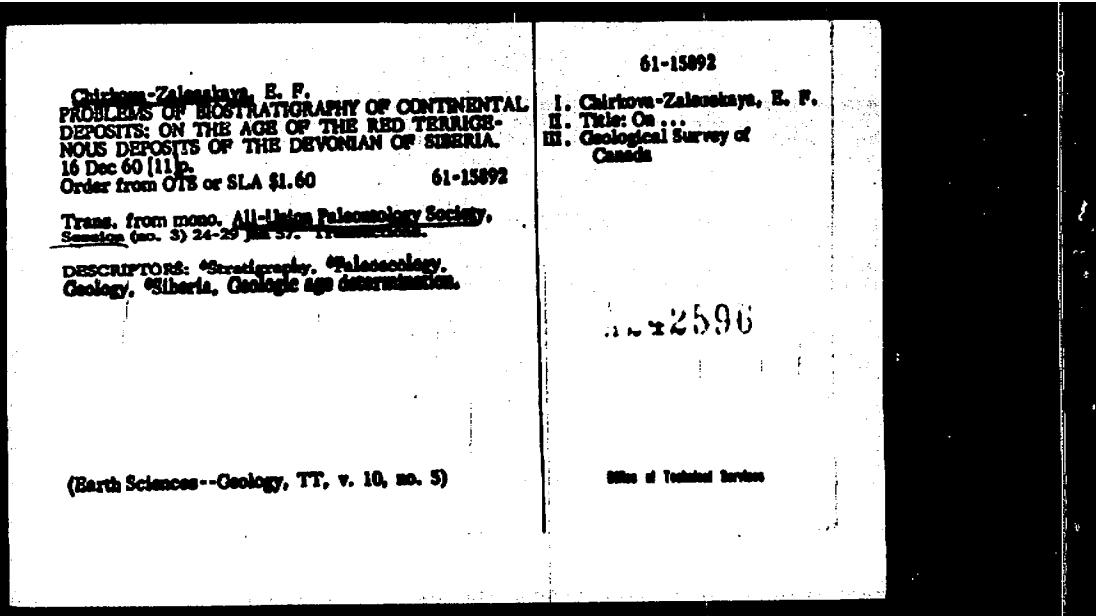
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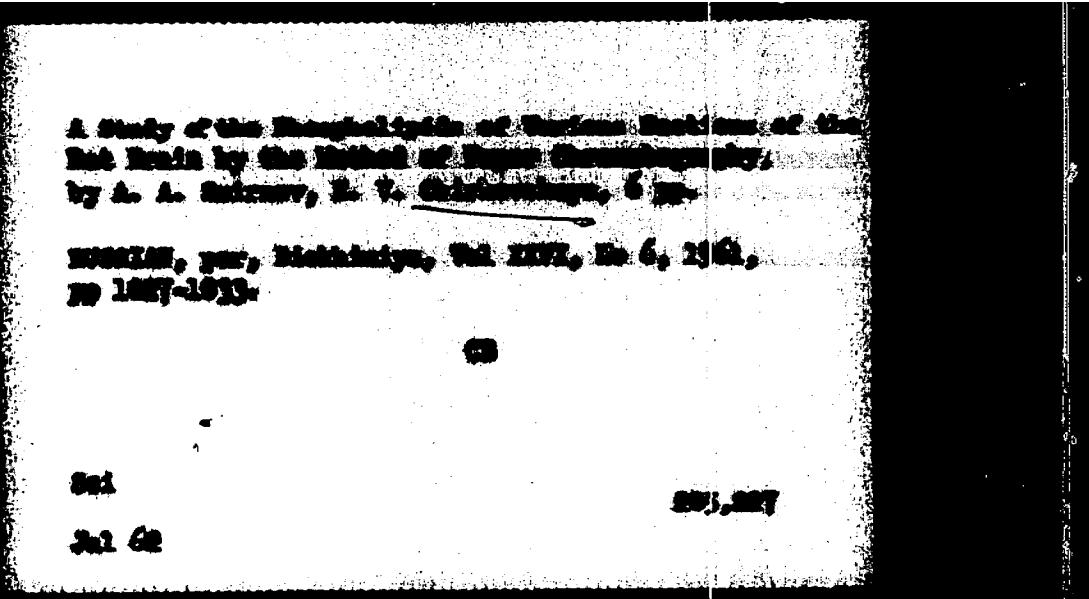
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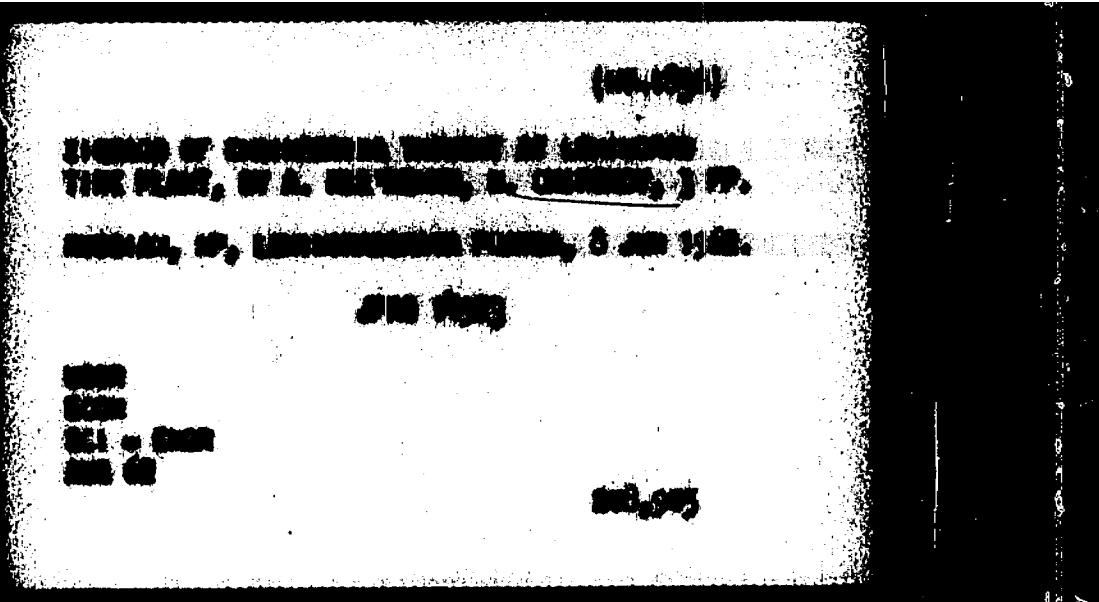
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